Development of hair-save unhairing method using organic thio compounds in pre-tanning stages of leather production.

Ariful Hai Quadery¹, Md. Tushar Uddin¹, Md. Abul Kashem Azad¹, Murshid Jaman Chowdhury¹, Amal Kanti Deb², Md. Nazmul Hassan³

¹Leather Research Institute, BCSIR, Nayarhat, Savar, Dhaka-1350, Bangladesh.

² Institute of Leather Engineering & Technology, University of Dhaka, 44-50 Hazaribugh, Dhaka-1209, Bangladesh.

³ Department of Applied Chemistry & Chemical Engineering, University of Dhaka, Dhaka-1000, Bangladesh.

Abstract

The function of liming and unhairing is to remove hair, interfibrillary components, fatty matters and epidermis and to open up fiber structure. The process of unhairing depends upon the phenomenon of destroying or loosening and removal of hair by chemical and mechanical means. The keratinous material (hair, hair root, epidermis etc.) are eliminated from the pelts conventionally with the mixture of sulphides (Na₂S, NaHS) and lime which contribute to the high Chemical Oxygen Demand (COD), Biological Oxygen Demand (BOD), Total Dissolved Solid (TDS) etc. in the tannery effluent. The study showed that significant amount of reduction in pollution concerning parameters like COD, BOD, suspended solids, Nitrogen and Sulphide in tannery waste water due to use of thio compound as unhairing agent. This reduction is explained by the fact that thio compound is non toxic. The intact hair is also collected as new raw material for fertilizer rather than being discharged with the effluent. The strength and organoleptic properties, chromium content and SEM analysis of the leather processed using thio compound, indicates that the quality of leather is also considerably improved.

This paper is focused on development of eco-friendly hair saving unhairing method using organic thio compound as an unhairing agent and to reduce the use of lime and sulphide in liming operation. It also discusses about the quality of leather and discharged liming waste water and compare with the conventional liming.

Keywords: Biological Oxygen Demand (BOD), Chemical oxygen demand (COD), Hair-save, Immunization of keratin, Liming, Scanning electron micrograph, Unhairing.

1 Introduction

HE leather processing involves various operations in a Tcascade manner form raw hide to crust leather. The complete leather manufacturing process is divided into three fundamental stages: beam house or pre-tanning, tanning and post tanning. The beam house stages comprise of soaking, liming, unhairing, deliming, bating and pickling.

Soaking [1], [2] is the first pre-tanning operation for treatment of hides and skins with water to clean and rehydrate as green condition. In this stage hides and skins are washed and soaked with surfactants and anti-microbial compounds before further processing.

The soaked hides and skins are treated with lime and

- Ariful Hai Quadery, Leather Research Institute, BCSIR, Nayarhat, Savar, Dhaka-1350, Bangladesh, PH-01123456789. E-mail: quadery.Iri@gmail.com
 Md. Nazmul Hassan, Department of Applied Chemistry and Chemical Engineering, University of Device Device 1000
- Md. Nazmul Hassan, Department of Applied Chemistry and Chemical Engineering, University of Dhaka, Dhaka-1000, Bangladesh, PH-+8801714391916. E-mail: nazmulek@gmail.com
- Murshid Jaman Chowdhury, Leather Research Institute, BCSIR, Nayarhat, Savar, Dhaka-1350, Bangladesh, PH-+8801718333335. E-mail: jaman.bcsir@gmail.com

sodium sulphide mixture which gives desired swelling of collagen structure to open up the fiber bundle. The quality of ultimate finished leather largely depends on this operation.

A common industrial practice of unhairing and liming is one step. Unhairing is the process of removing hair from the pelt without any damage to them. Once the hair shaft detached from the hide surface it is free to float in the bath and can be separated by filtration (hair saving) or chemically dissolved in the bath itself (destructive unhairing). The hair dissolution implies a much higher organic pollution of the waste water, whereas hair saving technologies needs a proper disposal of the recovered hair.

1.1 Sulphide unhairing and liming



Fig. 1. Action of unhairing agent during Na₂S liming system sodium sulfide and sodium sulphydrate used are sufficient

structure of the hair [2]. In this process the skins are introduced into a drum or paddle and sodium sulfide, sulphydrate and lime are added. The destruction of the hair results rapidly and within (half an hour) the hair loses most of its fiber structure. The hair destruction system is quite common in leather process for all type of production.

In alkaline solution the keratin –S-S- bond may be broken according to the following equation [1], [5].

 $R-S-S-R + 2Na_2S = 2R-SNa + Na_2S_2$

1.2 Hair saving unhairing

The hair saving process involves [6], [7] the introduction of enzymes with lime in a balanced system with careful attention to temperature and concentrations. Removal of hair takes place by hydrolysis of the soft proteins in the hair follicle. The thio-compounds have been used in various industrial applications. It has the capacity to cleavage [8] the cystine molecule of keratin protein.

Here the principle of liming by immunization [3], [9], [10] is generally been used. The soaked hides /skins are pretreated in the drum with an alkali like calcium hydrate. Under the influence of alkali the cystine of the hair transforms into lanthiamine, which can no longer hydrolyzed by reduction (immunization). The subsequent addition of sodium sulphide, for example causes a reduction only in unimmunized hair roots which leads to hair loosening. The comparatively well preserved hair can then be recovered by means of a screen [11], [12].



Fig. 2. Action of unhairing agent during a hair-saving system

1.3 The aim of this study

- Application of organic thio compound as an unhairing agent to reduce the use of lime and sulphide in liming operation.
- 2. To evaluate the wet blue and leather quality.
- 3. To study the effect of discharge liming water comparison with conventional liming.

2 Materials and Methods

2.1 Raw hide

In the preliminary trials 8 pieces of raw cow hides (each approximately 2 Kgs) were taken and soaked conventionally. The concentration of thio organic compound was varied over the range of 0.5-2.0 with 1% lime in the liming operation. All the chemicals percentages were based on salt weight and are commercial grade except the organic thiol compound which was prepared from reagent grades chemicals. In this work (no of) domestic cow sides weight ranges 6-7 Kgs were selected as raw materials.

2.2 Shrinkage temperature analysis

The shrinkage temperature (Ts) of samples, which is a measure of hydrothermal stability of leather, was determined [13] using a Theis shrinkage meter. Each value reported is an average of three experiments.

2.3 Analysis of hide substance content of leather samples

Total nitrogen contents of leather samples were measured [13] according to GB4689 using BUCHI-339 device. Hide substance contents of leather samples were calculated by multiplying the measured total nitrogen contents of leather samples by 5.62.

2.4 Mechanical properties of crust leather samples

Mechanical properties such as tensile strength, elongation, tear strength and grain crack strength were measured [13] according to standard procedures. Each value reported was an average of four (2 along the backbone, 2 across the backbone) measurements.

2.5 Analysis of waste water

Waste water was collected from dehairing process for the testing of total solids (TS), chemical oxygen demand (COD) and biochemical oxygen demand (BOD) and sulphide by standard method [9]. The results reported were average values of three experiments for each sample.

2.6 Analysis of total nitrogen in waste water

Total nitrogen contents of waste water collected from dehairing process to deliming process was measured [9] according to standard method.

2.7 Microscopic studies

Experimental as well as control leather surfaces were examined for variations in surface characteristics using FEI, Quanta 200 Scanning Electron Microscope [9], [14]. Samples from crust leather sides were directly cut into specimens with uniform thickness without any pretreatment.

2.8 Visual assessment

The experiment and control leathers were assessed for organoleptic properties by three experts.

3 Results and Discussion

Recipe 1: Conventional soak and lime process for wet salted hides (% based on wet salt weight.)

Process	%	Chemi	cal	Temp	erature	Time i	n minutes		рН
Dirt soak	150	Water		28ºC		30-60			
Drain									
Soak	120	Water		28ºC					
	0.5	Sodium carbor	nate						
	0.25	Non-ionic emu	Isifier						
	1.0	Enzymatic soal	king agent			240-360		9.5-10.5	
Drain									
Wash	100	Water		26ºC		10			
Drain									
Lime	100	Water		26ºC					
	1.0	Liming auxilia	ry						
	1.0	Sodium hyd 72%	rosulphide			45			
	1.0	Lime							
	1.0	Sodium sulphi	de 62%						
	0.2	Non-ionic emu	Isifier			60			
	50	water		26ºC					
	2.0	lime				30			
Run on autom	atic-stop 50 mins	s/run 10 mins for 12	2-14 hours						
Drain	•								
Wash	100	water	26ºC		15				
Drain									
Drain									
Wash	100	water	26ºC		10				
Drain									
Take out for fl	eshing	•							

Recipe 2: Hair saving and lime process using organic thio compound for wet salted hide (%based on wet salt weight.)

Process	%	Chem	ical	Temp.	Time in minutes	рН
Dirt wash	150	Water	Water		30-60	
Drain						
Soak	120	Water		28ºC		
	0.5		m carbonate			
	0.25		onic emulsifier			
	1.0	Enzyn	natic soaking agent		240-360	9.5-10.5
Drain						
Wash	100	Water		26ºC	10	
Drain						
Immunise	120	Water		26ºC		
	1.2	Thio o	rganic compound			
	1.5	lime			60	
Hair release	1.0		m hydrosulphide 72%			
	0.2	Non-io	onic emulsifier		20	
					20 stop	
					90 with screening c	of hair
Lime	30	water		26ºC		
	1.5	lime				
	0.5	Sodiu	m sulphide 62%		30	
Run overnight fo	or (Run 10	mins, stop	50 mins) for 12-14 hours			
Drain						
Wash	100		water		10	
Drain						
Wash	100		water		10	
Drain and take o	out for flesh	ing				

Table 1: Unhairing at different dosage of thio compound

	Condition of limed pelt		
To be detected lime thio compound	Unhairing degree	Scud	
dosage (% on salted weight)			
0.5	No loosening of hair	Present	
1.0	Moderate loosening of hair	Present	
1.5	Good loosening of hair	Presence is less	
2.0	Excellent Loosening of hair	Clear pelt	

Average value of two experiments

3.1 Preliminary unhairing

Preliminary unhairing experiments were performed on laboratory scale. Satisfactory unhairing occurred at 2% concentration. The result of unhairing trials are tabulated in table-1. Unhairing studies conducted at the pilot plant scale on three match sides that unhaired using thio compound (recipe 2.) were unhaired using thio compound with three corresponding matched hide sides for the control (recip 1.)

From the visual observation during unhairing it was quite comfortable to assess the extent of hair and scud removal.

The processes were conducted by stainless steel laboratory drums rotating in temperature controlled bath.

The half sides were processed to wet blue and samples were collected for shrinkage temperature and determination of chrome content. The remaining half sides then tanned to crust leather and samples were collected for physical testing.

In recipe-1, 3.5% sodium sulphide and 4% lime are used. A solution of sodium sulphide alone with admixture with lime is a strong unhairing agent. It is reported that when lime alone is used, it requires longer time to cause hair removal. Unhairing effect of sodium sulphide is maximum when SH and OH ions in the solution are present equal quantities. It is also reported only 0.6%Na₂S is required for a hair burn process. In practice much higher amount of sodium sulphide are added. The main reason for this is the fact. The rate of unhairing is on the concentration of sulphide ions.

 $2Na_2S + 2 H_2O = 2 NaHS + 2 NaOH$

 $2NaSH + Ca (OH)_2 = 2NaOH + Ca(SH)_2$

Here sodium sulphydrate act as an unhairing agent.

Total period in conventional liming was about 20-22 hours from the beginning. Next morning, drained away the liquor and run the drum 20 minutes for pulping out the hair. Then rinsed in water, unhaired and fleshed.

When the keratin protein in the hair cannot be chemically degraded and solubilized, it is said that hair is immunized and that such resistance results from the immunization phenomenon.

In recipe 2., the sodium hydrosulphide was used to replace the sodium sulphide partialy and this reduces the alkalinity available from the dissociation of sodium sulphide in water. Thio compound were used as reducing agent (HS-CH₂-CO₂-H).

The soaked hides were pretreated in the drum with a lime and thio compound. Under the influence of lime, the cystine of the hair transforms into lanthionine, which can no longer hydrolysed by reduction. In alkaline solution, the keratine –S-S- bond may be broken according to the following equation [4].

 $R-S-S-R + H_2O = R-SOH + R-SH$

The subsequent addition of sodium sulphide causes reduction only unimmunized hair roots, which leads to hair loosening. Total 1.5% sodium sulphide was used in this experimental trial.

At pH 9.5-10.5, the thio compound spread as far as the root of the hair and started to hydrolyse the pre-keratin of the hair root. The pH value rose to 12.0-12.5 with the addition of lime and the reducing activity of thio compound consequently increases. Under the influence of lime the immunization of the hair shaft was observed within 90 minutes.

At this point, sodium sulphydrate was added which further reduced and hydrolysed the pre-keratin. The well preserved hair was recovered by screening.

In this work sulphide consumption was less in the developed process (recipe 2.) than the conventional recipe 1. The procedure can be carried out in a drum or paddle and also possible to collect hair.

3.1 Standardization of unhairing process

8 pcs of raw cow hide (each approximately 2 Kgs) were soaked. The concentration of thio organic compound was varied over the range of 0.5-2.0 with 1% lime. All the chemical percentages were based on soaked weight. The unhairing efficiency for each trial is given in table 1.

Test	Thio-compound unhairing	Sulphide unhairing
% Increase in weight	55	45
% Cr ₂ O ₃	3.3	3.5
% Wrinkles	Less prominent	Prominent
Shrinkage	102ºC	100ºC
Temperature		

Table 2: Physical and chemica	l action on prepared wet blue
-------------------------------	-------------------------------

The hair-saving unhairing process was evaluated by measuring % weight change and the degree of wrinkling. The % Cr_2O_3 in the resulting wet blue leather was also determined as it is shown in the table 2. Both the shrinkage temperature and chrome content values are found considerably improved.

Test	Thio compound	Sulfide
	unhairing	unhairing
Tearing load	172	165
(N)		
Grain	7.6	8.3
distention (mm)		
Shrinkage	101	101
temperature		
(°C)		
Tensile strength	24.32	22.43
(N/mm²)		
Elongation (%)	35.52	39.51

Table 3: Physical-Mechanical properties of crust leather

Table 3 displays the physical tests, tear strength, tensile strength for both processed leather. The thio compound treated leather can be used for shoe upper since the value for tearing load and grain cracking was more than 120 N and 7.0 mm for distension.

The hair-saving unhairing process showed good physicalmechanical properties, comparable with those of the traditional. In fact the tear strength of the experimental leather was slightly higher as compared to corresponding control leather.

Table 4: Technical	properties of	crust leather
--------------------	---------------	---------------

Technical	Thio-compound	Sulphide
properties	unhairing	unhairing
Roundness	5	5
Fullness	5	4
Softness	5	4

The unhairing quality was evaluated by the appearance of the hair root on the leather by the roundness and softness after being staked.

The assessment results of the technical properties of hair saving unhairing system incomparision with the traditional sulphide unhairing system are reported in Table 4. A conventional scale of grades ranging from 1 (worst performance) to 5(least performance) has been used. It may be observed that the crust leather obtained by hair saving unhairing system and traditional sulphide unhairing system show quite similar technical properties.

Table 5: Comparison between the Polluting charge of (unhairing Exhaust Bath of thio-compound and sulfide)

Test parameter	Hair saving	Hair pulping	
	(thio compound)	(sulphide	
		unhairing)	
COD mg O ₂ /I	22,700	47,900	
BOD mg O ₂ /I	12000	31,000	
Ntk mg/I	1800	2560	
TSS g/l	4.7	6.2	
S ² mg/l	447	771	

The beam house phases, particularly the liming process, make up the most constituent part of total pollution produced due to their considerable contribution in the chemical and biochemical demand and total nitrogen and suspended solid.

Hair-save unhairing process has been applied that significantly reduce (25-30%) the BOD, COD and the total solids in the liming discharge. The result also shows the reduction of the sodium sulphide used with the consequent reduction of nitrogen. This is due to the presence of pollutants in low amounts of waste water in the case of developed hair-save unhairing system.

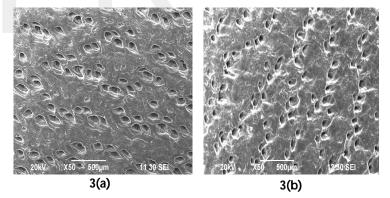


Figure: 3(a)- Scanning electron micrograph of a control and 3(b)- experimental crust leather showing the grain surface at magnification of X50

Scanning electron micrograph of crust leather samples processed through conventional unhairing and hair save unhairing process, showing the grain surface at a magnification of X50 are given in Fig. 3(a) and 3(b) respectively. It is seen that the hair pores of experimental samples appears to be much cleaner than control. This indicates that, hair save unhairing system able to perform unhairing without trace of hair. These observations are consistent with the conclusion mentioned above.

Conclusion

The possibility of unhairing using thio compound has been explored in this study. The use of thio compound has been found as excellent unhairing agent. The grain character and the strength properties of processed leather have been found to be good when composed to experimental leather. It is observed that hair was recovered easily by screening. This study was carried out in drum and the condition needs to modify for collection of hair. This brings about a significant reduction of the sodium sulphide and suspended solids with the consequent reduction in odour (H₂S gas).

The global concern for cleaner leather production has led tanneries to reduce the elimination of toxicity in their effluents. The possible application of hair saving unhairing process using thio compound confirmed.

Acknowledgments

It is great pleasure for us to offer our deepest thanks to all of respected teachers of the Department of Applied Chemistry and Chemical Engineering, University of Dhaka for expanding their helping hands whenever needed.

Thanks are also due to all the department staffs of the laboratory for their cooperation performing this research work and thanks to BCSIR (Bangladesh Council of Science and Industrial Research).

References

- A.D. Covington and T. Covington, *Tanning Chemistry: The science of leather*. Cambridge, UK: Royal Society of Chemistry, pp. 112-153, 2009.
- [2] F. O' Flaherty, W.T. Rody and R. Lollar, *The Chemisty and Technology of leather*, vol. 1. second Ed. ,R E K publishing company, Florida, pp. 229-322, 1978.
- [3] Ludvik, "Study on the scope for decrease of pollution load in leather processing," UNIDO Report, March 1997.
- [4] T.C. Thorstensen, *Practical Leather Technology*. Fourth Edition, R. E. Krieger Pub. Co., Florida, pp. 84-106, 1993.
- [5] E. Heidemann, *Fundamentals of leather manufacturing*, E. Roether KG, pp. 180-193, 1993.
- [6] H.P. Germann, "The evolution of the unhairing process as influenced by technological, economic, and ecological considerations," J. Amer. Leather Chem. Ass., vol. 92, pp. 84-92, 1997.
- [7] R.G. Paul, I. Mohammed, D. Davighi, V.L. Addy and A.D. Covington, "The use of neutral protease in

enzymatic unhairing," J. Amer. Leather Chem. Ass., vol. 96, pp.180-185, 2001.

- [8] A.M.S. Kamal, Y. Nomura, Y. Ishii and K. Shirai, "Properties of Bovine hair Keratins solubilized with thioglycolate," *J. Amer. Leather Chem. Ass.*, vol. 93, pp. 272-282, 1998.
- [9] C.S. Cantera, A.R. Angelinetti, G. Altobelli and G. Gaita, Hair saving enzyme assisted unhairing, "Influence of enzymatic products upon final leather quality," *J. soc. Leather Technol. Chem.*, vol. 80, pp. 83-86, 1996.
- [10] A. Masal, J.M. Morera, E. Bartoli and M. Borras, "Study on an unhairing process with hydrogen peroxide and Amines," J. Amer. Leather Chem. Ass., 95: pp.1-10, 2000.
- [11] D.G. Bailey, World Leather, vol. 8, no. 3, pp. 43,1995.
- [12] R.W. Cranston, M.H. Davis and JG. Scroggie, J. soc. Leather Technol. Chem., vol. 70, pp. 53, 1986.
- [13] SLTC, Official methods of analysis. UK, 1996.
- [14] P. Echlin, Scanning electron microscopy. Vol. 4, Hey wood, V.H.Ed., Academic press, pp. 307, 1971.

